

Stpm Chemistry Past Year Papers

STPM Chemistry (Semester 2)

## STPM CHEMISTRY SEMESTER 2 PRE-EXAM PRACTICE

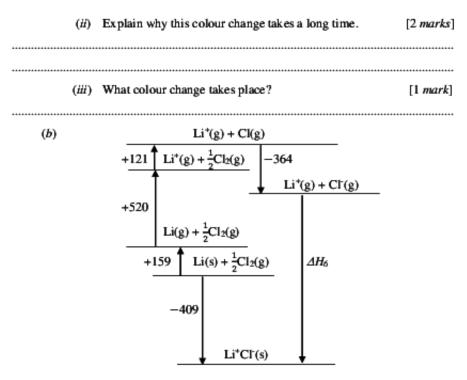
## Section A: Structured Questions Answer all questions.

1 (a) Chrome yellow has been used for a long time as a yellow pigment in oil paintings.

$SO_4^{2-}(aq) + 4H^+(aq) + 2e \rightleftharpoons 2H_2O(l) + SO_2(g)$	$E^{o} = +0.17 \text{ V}$
$CrO_4^{2-}(aq) + 8H^*(aq) + 3e \rightleftharpoons 4H_2O(l) + Cr^{3+}(aq)$	$E^{o} = +1.33 \text{ V}$

(i) By using the standard electrode potentials above, explain why the yellow colour of the painting exposed to an atmosphere contains sulphur dioxide.

[2 marks]



- 1 -

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